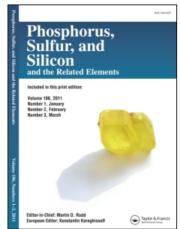
This article was downloaded by:

On: 27 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

Synthesis and Characterization of Novel Cycloalkanespiro-5-Hydantoin Phosphonic Acids

Emilia D. Naydenova^a; Petar T. Todorov^a; Kolio D. Troev^b

^a University of Chemical Technology and Metallurgy, Department of Organic Chemistry, Sofia, Bulgaria ^b Institute of Polymers, Bulgarian Academy of Sciences, Sofia, Bulgaria

Online publication date: 02 July 2010

To cite this Article Naydenova, Emilia D. , Todorov, Petar T. and Troev, Kolio D.(2010) 'Synthesis and Characterization of Novel Cycloalkanespiro-5-Hydantoin Phosphonic Acids', Phosphorus, Sulfur, and Silicon and the Related Elements, 185: 7,1315-1320

To link to this Article: DOI: 10.1080/10426501003751254 URL: http://dx.doi.org/10.1080/10426501003751254

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Phosphorus, Sulfur, and Silicon, 185:1315-1320, 2010

Copyright © Taylor & Francis Group, LLC ISSN: 1042-6507 print / 1563-5325 online DOI: 10.1080/10426501003751254



SYNTHESIS AND CHARACTERIZATION OF NOVEL CYCLOALKANESPIRO-5-HYDANTOIN PHOSPHONIC ACIDS

Emilia D. Naydenova, Petar T. Todorov, and Kolio D. Troev²

¹University of Chemical Technology and Metallurgy, Department of Organic Chemistry, Sofia, Bulgaria

²Institute of Polymers, Bulgarian Academy of Sciences, Sofia, Bulgaria

Keywords α -Aminophosphonic acids; cycloalkanespiro-5-hydantoins; phosphorus trichloride

INTRODUCTION

 α -Aminophosphonic acids are structurally analogous to aminocarboxylic acids. The structures of aminophosphonic acids are of interest because of the biological role of their derivatives. These compounds possess a variety of biological activities, including anticancer, antiviral, antibacterial, antifungal, and other activities. The established antiproliferative effects, together with the low mammalian toxicity of these agents, has conditioned tremendous interest towards designing novel antineoplastic agents. Synthesis of aminophosphonic acids is an active area of research, and many methods are now available. Novel α -aminophosphonic acids with moderate clastogenic effect were synthesized reacting 1,3-oxazolidin-2-one derivatives with formaldehyde and phosphorus trichloride. Novel α -aminophosphonic acids or heterocyclic rings, introduced into the molecular skeleton, increase its rigidity and modify the electronic effects. Thus in recent years, many cyclic α -aminophosphonic acids or aminophosphonates have been prepared. A series of α , adisubstituted cyclic derivatives of N-(phosphonomethyl)glycine were obtained from cycloalkaneaminocarboxylic acids, and their biological activity was studied.

Hydantoin derivatives are synthetically valuable, e.g., as precursors to α -amino acid and pyruvic acid derivatives. ^{17–19} Hydantoins, substituted at C-5, are important medicinal

Received 14 July 2008; accepted 5 March 2010.

We gratefully acknowledge the financial support by the University of Chemical Technology and Metallurgy-contract No. 10507.

Address correspondence to Prof. Dr. Kolio D. Troev, Institute of Polymers, Bulgarian Academy of Sciences, Sofia 1113, Bulgaria. E-mail: ktroev@polymer.bas.bg

compounds. Numerous applications have been found for hydantoin derivatives due to their antidepressant²⁰ and antiviral activities,²¹ the inhibition of binding of HIV to lymphocytes,²² as well as their anticonvulsant and cardiac antiarrhythmic effects.²³ Taking into account this fact, we describe in this article the synthesis of novel α -aminophosphonic acids with potential biological activity by reacting phosphorus trichloride with formaldehyde and cycloalkanespiro-5-hydantoins.

RESULTS AND DISCUSSION

In our present studies, we used the modified procedure described by Engelmann and Pikl, 24 according to which phosphorus trichloride, formaldehyde, and oxazolidinone derivatives are used as the starting compounds for the preparation of the aminophosphonic acids. The new α -aminophosphonic acids were obtained from cycloalkanespiro-5-hydantoins, formaldehyde, and phosphorus trichloride as shown in Scheme 1. Cycloalkanespiro-5-

Scheme 1

hydantoins **1a–1d**, phosphorus trichloride, and formaldehyde were used as starting compounds in a molar ratio of 1:2:2. The compounds **1a–1d** were prepared by the Bucherer–Lieb reaction. The rate of hydroxymethylation of the two nitrogen atoms is different. Hydroxymethylation at the N-3 atom proceeds faster because of the higher acidity of the NH-bond compared to that of the N-1 atom. In the 31 P NMR spectrum of the reaction product, two signals are observed at 20.6 ppm (triplet with $^{2}J_{PH} = 13.5$ Hz) and at 18.1 ppm (triplet with $^{2}J_{PH} = 11.0$ Hz). The signal at 20.6 ppm can be assigned to the phosphorus atom of the fragment connected to N-3, and that at 18.1 ppm to the phosphorus atom of the fragment connected to N-1. The intensity ratio of 2:1 confirms the difference in the reactivity of the two NH-groups in the cycloalkanespiro-5-hydantoins (see Table I).

Unfortunately, the new synthesized compounds could not be separated. The formation of [(2,4-dioxo-1,3-diazaspiro-alkane-3-yl)methyl]phosphonic acids **2a–2d** and [(2,4-dioxo-1,3-diazaspiro-alkane-1,3-diyl)dimethyl]diphosphonic acids **3a–3d** was confirmed by ¹H, ¹³C{¹H}, ³¹P NMR, and IR spectroscopy.

	δ^{31} P (ppm) / $^2J_{PH}$ (Hz)
2a	20.6 (t) / 13.5
3a	18.1 (t) / 11.0
2b	19.4 (t) / 15.0
3b	15.5 (t) / 12.0
2c	20.0 (t) / 11.6
3c	17.0 (t) / 12.2
2d	21.1 (t) / 14.5
3d	17.6 (t) / 12.0

Table I ³¹P NMR data of the new cycloalkanespiro-5-hydantoin phosphonic acids 2 and 3

In the ^1H NMR spectra of all compounds **3a–3d**, doublets with a characteristic $^2J_{\text{PH}}$ coupling constant of 11–15 Hz are observed for the P–CH₂ protons. Direct evidence for the formation of the P–C bond in the P–CH₂ fragment is the doublets that appear in the $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of all compounds **3a–3d** with a characteristic $^1J_{\text{PC}}$ coupling constant of 154–156 Hz (see the Experimental section). In the ^{31}P NMR spectrum of the reaction mixture, an additional signal appears at 22 to 23 ppm, which is characteristic of a phosphonic structure. Further studies to elucidate the structure of these products and their formation are in progress.

CONCLUSION

In this article, the synthesis and characterization of new [(2,4-dioxo-1,3-diazaspiro-alkane-3-yl)methyl]phosphonic acids and [(2,4-dioxo-1,3-diazaspiro-alkane-1,3-diyl)dimethyl] diphosphonic acids, prepared by reacting cycloalkanespiro-5-hydantoins with formaldehyde and phosphorus trichloride, is described.

EXPERIMENTAL

Instruments and Reagents

Cycloalkanones, phosphorus trichloride, paraformaldehyde, and the solvents were purchased from Fluka and Merck, and were used without further purification.

The infrared (IR) spectra were recorded in KBr pellets with a Perkin-Elmer Model 1600 Series FT-IR instrument. Melting points (mp) were determined on a Koffler microscope and were uncorrected. The purity of the products was checked by TLC on precoated plates of silica gel 60 F254 (Merck) using as the mobile phase a 3:1:1 mixture of n-BuOH, AcOH, and H₂O. Spots on TLS chromatograms were detected by chlorine/o-tolidine reaction. The ¹H, ¹³C, and ³¹P NMR spectra were obtained with a Bruker DRX 400 spectrometer at 400.13, 100.61, and 161.97 MHz, respectively. ¹³C NMR spectra were fully ¹H decoupled. Chemical shifts are reported in δ values (ppm), and J values are reported in hertz (Hz).

The compounds 1a-1d were prepared by the Bucherer-Lieb reaction. 25,27

Preparation of Cycloalkanespiro-5-hydantoins (1a–1d) by Bucherer–Lieb Synthesis

To a solution of the cycloalkanone (0.0594 mol) in ethanol (55 mL) and water (50 mL), sodium cyanide (6.63 g, 0.1 mol) and ammonium carbonate (34.6 g, 0.36 mol) were

added. The mixture was refluxed for 6 h with stirring. After dilution with water, the cooled mixture was acidified with concentrated hydrochloric acid. The crude cycloalkanespiro-5-hydantoin precipitated overnight upon cooling at 5°C. The pure compound was crystallized from water as colorless crystals.

- **1,3-Diazaspiro[4.4]nonane-2,4-dione (1a).** Yield 96.0% (8.79 g), $R_f = 0.74$, mp = 204–205°C; IR (KBr, cm $^{-1}$): 3215 ($^{3}N-H$), 3074 ($^{1}N-H$), 1788 ($^{2}C=O$), 1737 ($^{4}C=O$). ^{1}H NMR (d 6 -DMSO): $\delta = 1.8-2.1$ (m, 8H, CH $_{2}$), 7.3 (s, 1H, NH), 10.4 (s, 1H, NH). $^{13}C\{^{1}H\}$ NMR (d 6 -DMSO): $\delta = 23.8$ (C-3,4), 35.8 (C-2,5), 69.9 ($^{-}C-$), 157.4 2-C=O, 168.9 4-C=O.
- **1,3-Diazaspiro[4.5]decane-2,4-dione (1b).** Yield 88.2% (8.81 g), $R_f = 0.87$, mp = 219–220°C; IR (KBr, cm $^{-1}$): 3198 (^{3}N –H), 3070 (^{1}N –H), 2951–2858 (C–H), 1775 (^{2}C =O), 1727 (^{4}C =O). ^{1}H NMR (d 6 -DMSO): $\delta = 1.6$ –1.9 (m, 10H, C \underline{H}_2), 8.2 (s, 1H, NH), 10.3 (s, 1H, NH). $^{13}C\{^{1}H\}$ NMR (d 6 -DMSO): $\delta = 23.1$ (C-3,5), 25.4 (C-4), 36.0 (C-2,6), 68.7 (–C–), 157.0 2-C=O, 167.7 4-C=O.
- **1,3-Diazaspiro[4.6]undecane-2,4-dione (1c).** Yield 76.2% (8.23 g), $R_f = 0.87$, mp = 213–215°C; IR (KBr, cm⁻¹): 3282 (³N–H), 3055 (¹N–H), 2941–2852 (C–H), 2347 (NH), 1770 (²C=O), 1709 (⁴C=O). ¹H NMR (d⁶-DMSO): $\delta = 1.5-1.8$ (m, 12H, C \underline{H}_2), 8.2 (s, 1H, NH), 10.4 (s, 1H, NH). ${}^{13}C\{{}^{1}H\}$ NMR (d⁶-DMSO): $\delta = 21.9$ (C-3,6), 26.8 (C-4,5), 35.9 (C-2,7), 66.7 (–C–), 157.3 2-C=O, 172.7 4-C=O.
- **1,3-Diazaspiro[4.7]dodecane-2,4-dione (1d).** Yield 91.0% (10.59 g), $R_f = 0.79$, mp = 240–241°C; IR (KBr, cm⁻¹): 3250 (3 N–H), 3183 (1 N–H), 1765 (2 C=O), 1713 (4 C=O). 1 H NMR (d 6 -DMSO): $\delta = 1.5$ –2.1 (m, 14H, C \underline{H}_2), 8.2 (s, 1H, NH),10.4 (s, 1H, NH). 13 C{ 1 H} NMR (d 6 -DMSO): $\delta = 20.0$ (C-3,7), 24.2 (C-5), 26.8 (C-4,6), 38.1 (C-2,8), 68.0 (–C–), 158.1 2-C=O, 176.0 4-C=O.

Preparation of [(2,4-Dioxo-1,3-diazaspiro-alkane-3-yl)methyl]phosphonic Acids (2a–2d), [(2,4-Dioxo-1,3-diazaspiro-alkane-1,3-diyl) dimethyl]diphosphonic Acids (3a–3d) in Ratio of 2:1: General Procedure

The respective cycloalkanespiro-5-hydantoins (0.0059 mol) and paraformaldehyde (0.36 g, 0.0119 mol) were placed under argon in a four-necked round-bottomed flask equipped with a magnetic stirrer, reflux condenser, thermometer, dropping funnel, and argon inert. Under vigorous stirring glacial acetic acid (5.95 mL) was added dropwise. A white suspension formed. The reaction mixture was refluxed (\sim 115°C) for 12.5 h, after which it became a clear solution. Then the temperature was lowered to 20°C and 1.04 mL (1.63 g, 0.0119 mol) of phosphorus trichloride was added dropwise. During and after the addition hydrogen chloride evolved. The reaction mixture was refluxed (\sim 118°C) for 6 h. Subsequently 6.80 mL of distilled water was added. After 4 h further refluxing, the reaction mixture was concentrated under reduced pressure.

The solvent was evaporated, and the product left was dissolved in a large excess of methanol. After filtration, the filtrate was evaporated to give a residue, which was purified by precipitation in methanol/ethyl acetate and collected by filtration. The purification step was repeated several times to give compounds (2a-2d and 3a-3d).

[(2,4-Dioxo-1,3-diazaspiro[4.4]nonane-3-yl)methyl]phosphonic Acid (2a), [(2,4-Dioxo-1,3-diazaspiro[4.4]nonane-1,3-diyl)dimethyl]diphosphonic Acid (3a). Yellow oil, yield 45.0% (0.66 g), $R_f = 0.45$; IR (KBr, cm⁻¹): 3424 (OH), 2942–2854 (C-H), 1730 (^2C=O), 1708 (^4C=O), 1156 (P=O), 1114, 1012 (P-O-H). \(^1\)H NMR (400.13 MHz, CD₃OD): $\delta = 1.1$ —1.9 (m, 8H, CH₂), 3.2 (d, $^2J_{PH} = 11.5$ Hz, 2H, P-CH₂),

3.3 (d, ${}^2J_{\text{PH}} = 10.0$ Hz, 2H, P $-\text{CH}_2$), 5.6 (br.s., 2H, P-OH), 8.2 (br.s., 2H, P-OH). ${}^{13}\text{C}\{{}^{1}\text{H}\}$ NMR (100.61 MHz, CD₃OD): $\delta = 21.5$ (CH₂), 24.2 (CH₂), 25.2 (CH₂), 34.5 (d, ${}^{1}J_{\text{PC}} = 149.6$ Hz, P $-\text{CH}_2$), 59.0 (d, ${}^{1}J_{\text{PC}} = 154.1$ Hz, P $-\text{CH}_2$), 63.4 (-C-), 156.4 (2-C=O), 177.7 (4-C=O). ${}^{31}\text{P}$ NMR (161.97 MHz, CD₃OD): = 18.1 (t, ${}^{2}J_{\text{PH}} = 11.0$ Hz), 20.6 (t, ${}^{2}J_{\text{PH}} = 13.5$ Hz) intensity integral ratio of 2:1.

[(2,4-Dioxo-1,3-diazaspiro[4.5]decane -3-yl)methyl]phosphonic Acid (2b), [(2,4-Dioxo-1,3-diazaspiro[4.5]decane-1,3-diyl)dimethyl]diphosphonic Acid (3b). Yellow oil, yield 48.0% (0.74 g), R_f = 0.41; IR (KBr, cm⁻¹): 3397 (OH), 2942–2854 (C−H), 1740 (²C=O), 1713 (⁴C=O), 1150 (P=O), 1036, 914 (P−O−H). ¹H NMR (400.13 MHz, CD₃OD): δ = 1.5—2.1 (m, 10H, CH₂), 3.5 (d, ² J_{PH} = 11.0 Hz, 2H, P−CH₂), 3.8 (d, ² J_{PH} = 9.0 Hz, 2H, P−CH₂), 5.7 (br.s., 2H, P−OH), 8.1 (br.s., 2H, P−OH). ¹³C{¹H} NMR (100.61 MHz, CD₃OD): δ = 21.0 (C-4), 22.0 (C-3,5), 26.0 (C-2,6), 38.5 (d, ¹ J_{PC} = 150.5 Hz, P−CH₂), 60.0 (d, ¹ J_{PC} = 156.6 Hz, P−CH₂), 68.0 (−C−), 156.7 (2-C=O), 176.7 (4-C=O). ³¹P NMR (161.97 MHz, CD₃OD): δ = 15.5 (t, ² J_{PH} = 12.0 Hz), 19.4 (t, ² J_{PH} = 15.0 Hz) intensity integral ratio of 2:1.

[(2,4-Dioxo-1,3-diazaspiro[4.6]undecane-3-yl)methyl]phosphonic Acid (2c), [(2,4-Dioxo-1,3-diazaspiro[4.6]undecane-1,3-diyl)dimethyl]diphosphonic Acid (3c). Yellow oil, yield 42.0% (0.68 g), R_f = 0.43; IR (KBr, cm⁻¹): 3348 (OH), 2955–2859 (C—H), 1745 (²C=O), 1710 (⁴C=O), 1171 (P=O), 1013, 965 (P—O—H). ¹H NMR (400.13 MHz, CD₃OD): δ = 1.2—1.9 (m, 12H, CH₂), 3.3 (d, ²J_{PH} = 12.0 Hz, 2H, P—CH₂), 3.5 (d, ²J_{PH} = 11.0 Hz, 2H, P—CH₂), 5.2 (br.s., 2H, P—OH), 8.4 (br.s., 2H, P—OH). ¹³C{¹H} NMR (100.61 MHz, CD₃OD): δ = 20.0 (CH₂), 24.9 (CH₂), 26.0 (CH₂), 34.0 (CH₂), 37.0 (d, ¹J_{PC} = 156.6 Hz, P—CH₂), 57.6 (d, ¹J_{PC} = 152.1 Hz, P—CH₂), 61.0 (–C—), 159.7 (2-C=O), 176.0 (4-C=O). ³¹P NMR (161.97 MHz, CD₃OD): δ = 17.0 (t, ²J_{PH} = 12.2 Hz), 20.0 (t, ²J_{PH} = 11.6 Hz) intensity integral ratio of 2:1.

[(2,4-Dioxo-1,3-diazaspiro[4.7]dodecane-3-yl)methyl]phosphonic Acid (2d), [(2,4-Dioxo-1,3-diazaspiro[4.7]dodecane-1,3-diyl)dimethyl]diphosphonic Acid (3d). Yellow oil, yield 45.8% (0.79 g), R_f = 0.47; IR (KBr, cm⁻¹): 3261 (OH), 2925–2855 (C−H), 1765 (2 C=O), 1710 (4 C=O), 1180 (P=O), 1012, 969 (P−O−H). 1 H NMR (400.13 MHz, CD₃OD): δ = 1.1—1.9 (m, 8H, CH₂), 3.2 (d, 2 J_{PH} = 11.5 Hz, 2H, P−CH₂), 3.3 (d, 2 J_{PH} = 10.0 Hz, 2H, P−CH₂), 5.8 (br.s., 2H, P−OH), 8.0 (br.s., 2H, P−OH). 13 C{ 1 H} NMR (100.61 MHz, CD₃OD): δ = 21.8 (C-5), 24.4 (C-4,6), 27.7 (C-3,7), 33.0 (C-2,8), 38.5 (d, 1 J_{PC} = 153.2 Hz, P−CH₂), 58.7 (d, 1 J_{PC} = 155.5 Hz, P−CH₂), 64.6 (−C−), 155.9 (2-C=O), 178.9 (4-C=O). 31 P NMR (161.97 MHz, CD₃OD): δ = 17.6 (t, 2 J_{PH} = 12.0 Hz), 21.1 (t, 2 J_{PH} = 14.5 Hz) intensity integral ratio of 2:1.

REFERENCES

- (a) K. D. Troev, Chemistry and Application of H-Phosphonates (Elsevier, Amsterdam 2006);
 (b) P. Kafarski and B. Lejczak, Aminophosphonic and Aminophosphinic Acids: Chemistry and Biological Activity (Wiley, New York, 2000), p. 407.
- 2. E. W. Logusch, D. M. Walker, J. F. McDonald, J. E. Franz, J. J. Villafranca, C. L. Dilanni, J. A. Colanduoni, B. Li, and J. B. Shineller, *Biochemistry*, **29**, 366 (1990).
- 3. T. D. Meek and J. J. Villiafranca, Biochemistry, 19, 5513 (1980).
- 4. J. Oleksyszyn, B. Boduszek, Ch. M. Kam, and J. C. Powers, J. Med. Chem., 37, 226 (1994).
- 5. W. W. Smith and P. A. Bartlett, J. Am. Chem. Soc., 120, 4622 (1998) and references cited therein.
- 6. J. W. Goding, *Monoclonal Antibodies: Principles and Practice* (Academic Press, New York, 1986).

- 7. M. Kabachnik and T. Medved, Dokl. Acad. Nauk SSSR, 83, 689 (1952).
- 8. E. K. Fields, J. Am. Chem. Soc., 74, 1528 (1952).
- 9. I. A. Natchev, Liebigs. Ann. Chem., 9, 861 (1988).
- (a) K. Troev, *Phosphorus, Sulfur, and Silicon*, 127, 167 (1997) (b) K. Troev, Sh. Cremer, and G. Haegele, *Heteroatom Chem.*, 10, 627 (1999) (c) E. Naydenova, K. Troev, M. Topashka-Ancheva, G. Haegele, I. Ivanov, and A. Kril, *Amino Acids*, 33, 695 (2007).
- 11. P. P. Giannousis and P. A. Bartlett, J. Med. Chem., 30, 1603 (1987).
- 12. J. Huang and R. Chen, Heteroatom Chem., 11, 480 (2000).
- E. Naydenova, M. Topashka-Ancheva, P. Todorov, Ts. Yordanova, and K. Troev, *Bioorg. Med. Chem.*, 14, 2190 (2006).
- P. Todorov, E. Naydenova, R. Petrova, B. Shivachev, and K. Troev, *Acta Crystallogr. Sect. C*, 62, o661 (2006).
- (a) N. Rabasso, N. Louaisil, and A. Fadel, *Tetrahedron*, 62, 7445 (2006) (b) N. Louaisil, N. Rabasso, and A. Fadel, *Synthesis*, 289 (2007).
- E. Naydenova, P. Todorov, M. Topashka-Ancheva, G. Momekov, Ts. Yordanova, S. Konstantinov, and K. Troev, Eur. J. Med. Chem., 43, 1199 (2008).
- N. A. Meanwell, H. R. Roth, E. C. R. Smith, D. L. Wedding, and J. J. K. Wright, *J. Org. Chem.*, 56, 6897 (1991).
- 18. R. Sarges and P. J. Oates, Prog. Drug. Res., 40, 99 (1993).
- P. Barraclough, M. L. Bolofo, H. Giles, J. Gillam, C. J. Harris, M. G. Kelly, P. Leff, A. McNeill,
 A. D. Robertson, R. J. Stepney, and B. J. R. Whittle, *Bioorg. Med. Chem.*, 4, 81 (1996).
- F. L. Wessels, T. J. Schawn, and S. F. Pong, J. Pharm. Sci., 69, 1102 (1980).
- 21. A. A. El-Barbarly, A. I. Khodair, E. B. Pedersen, and C. Nielsen, J. Med. Chem., 37, 73 (1994).
- 22. W. M. Cloyd, S. W. Lynn, K. Ramsey, and S. Baron, Virology, 173, 581 (1989).
- 23. M. L. Brown, G. B. Brown, and W. J. Brouillette, J. Med. Chem., 40, 602 (1997).
- 24. (a) M. Engelmann and J. Pikl, U.S. Pat. 2,304,156 (1942); (b) J. Pikl, U.S. Pat. 2,328,358 (1943).
- 25. H. Bucherer and V. Lieb, J. Prakt. Chem., 141, 5 (1934).
- (a) J. Rubido, C. Pedregal, M. Espada, and J. Elguero, *Synthesis*, 3, 307 (1985) (b) E. Ware, *Chem. Reviews*, 46, 403 (1950).
- E. Naydenova, N. Pencheva, J. Popova, N. Stoyanov, M. Lazarova, and B. Aleksiev, *Il Farmaco*, 57, 189 (2002).